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 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ fast
 $\text{O} + \text{O}_3 \rightarrow \text{O}_2 + \text{O}_2$ slow
 a write the overall balanced chemical equation
 $2 \text{O}_3 \rightarrow 3 \text{O}_2$ b identify any intermediates within the mechanism
 c what is the order with respect

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