

IGCSE Physics

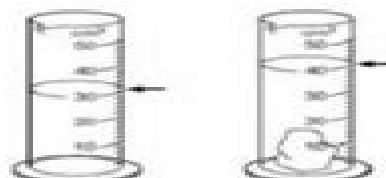
Volume, Density and Mass

Volume (cm^3) – the amount of space an object occupies

How to measure volume:

- **For Liquids** – use measuring cylinder and read the volume directly
- **For Regular Solids** – calculate using the objects measurements:
 - Cuboid: height \times width \times length
 - Sphere: $\frac{4}{3} \times \pi \times \text{radius}^3$.
 - Cylinder: $\pi \times \text{radius}^2 \times \text{height}$
- **For Irregular Solids** – use the displacement method

Displacement method – place a selection of water in a measuring cylinder and measure the volume. Then add the object into the water and measure it again. The volume of the object is the difference in volumes. *E.g.*



Initial Volume = 30cm^3
 Final Volume = 40cm^3
 $40\text{cm}^3 - 30\text{cm}^3 = 10\text{cm}^3$
 Object Volume = 10cm^3

Density (g/cm^3) – mass per unit of volume

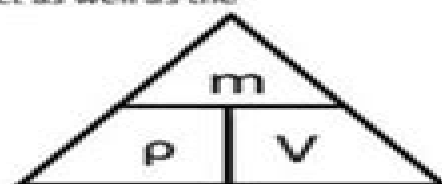
Relative Density – the ratio between the density of that substance and the density of water. It has no limit.

Mass (g) – a measure of the amount of matter present in an object as well as the characteristic which resists change in motion

The calculation of volume is $\text{volume} = \frac{\text{mass}}{\text{density}}$

The calculation of density is $\text{density} = \frac{\text{mass}}{\text{volume}}$

The calculation of mass is $\text{mass} = \text{density} \times \text{volume}$



Density of air – 1.225 kg/m^3

Density of water – 1000 kg/m^3

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