

Lab 8

Cellular Respiration



Section 1 – Review of Energy Molecules

[2] Welcome to this week's lab on cellular respiration. This lab continues the exploration of how organisms obtain and use energy. As you study this diagram you will see that photosynthesis and cellular respiration are basically opposite reactions. In photosynthesis, autotrophs capture light energy and store the energy in glucose. In cellular respiration, organisms break down glucose to release energy.

[3] Before getting to respiration, let's review some concepts from last week. Look at Section 1 and see if you can answer the first two questions and then return to the program to see if you remembered correctly.

[4] How did you do? Did you remember that ATP carries energy in a chemical bond and NADPH carries energy in high energy electrons? If so great! If not, you can look back at Section 2 of Lab 7 for a brief review.

[5] Cellular respiration uses two new molecules to carry high energy electrons. These molecules are seen here. Look at them and see if you can recognize which of these molecules have picked up the high energy electrons.

[6] OK, did you figure it out? See if you did by answering this question.

[7] Did you remember that the negative charge of electrons is cancelled by the positive charge of hydrogen ions and thus conclude that the NADH and the FADH_2 are the molecules that contain the high energy electrons? With this information you can answer the last question in Section 1 and then return to the program.

Section 2 – Biochemical Pathways in Cellular Respiration

[8] Now we will continue our exploration of energy and cellular respiration. All organisms, including plants, break down organic molecules to obtain energy. This energy is then used to produce ATP. The ATP in turn is used to carry on cellular activities.

[9] Just like we saw with photosynthesis, there are several different chemical pathways and subreactions associated with how organisms release energy from organic molecules to produce ATP. The two main pathways are respiration and fermentation. List these pathways in the first part of Section 2.

[10] Last week we saw how plants produce glucose using the energy and electrons from ATP and NADPH. Do you remember where the energy is stored in glucose?

[11] Yes, the energy is stored in both chemical bonds and high energy electrons. The chemical reactions involved in extracting energy from glucose will use the energy released when the bonds in glucose are broken and the high energy electrons are released.

Lab Manuals For Pierce College

Thomas Griffiths



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Forensic Anthropology Laboratory Manual Steven N. Byers, Chelsey A. Juarez, 2016-10-04 This manual provides students in academic laboratory courses with hands on experience of the major processes of forensic anthropology Designed to accompany the textbook Introduction to Forensic Anthropology the manual introduces core procedures and protocol with exercise worksheets to reinforce the methodologies of forensic anthropology and enhance student comprehension For the fourth edition the manual has been updated in line with the textbook incorporating new methods figures and worksheets Each chapter contains explanations of the terminology osteological features and measurements needed to understand each of the topics Chapters may be covered in one session or multiple sessions and include lists of both basic and optional lab materials enabling instructors to tailor each lab to the resources they have available Transistor Laboratory Manual Milton Sol Kiver, Bernard Van Emden, 1962 A Laboratory Manual Ira Remsen, 1902 *Laboratory Manual for Exercise Physiology* G. Gregory Haff, Charles Dumke, 2021-06 Laboratory Manual for Exercise Physiology Second Edition With HKPropel Access provides guided opportunities for students to translate their scientific understanding of exercise physiology into practical applications in a variety of settings Written by experts G Gregory Haff and Charles Dumke the text builds upon the success of the first edition with full color images and the addition of several new online interactive lab activities The revitalized second edition comprises 16 laboratory chapters that offer a total of 49 lab activities Each laboratory chapter provides a complete lesson including objectives definitions of key terms and background information that sets the stage for learning Each lab activity supplies step by step procedures providing guidance for those new to lab settings so that they may complete the procedures New features and updates in this edition include the following Related online learning tools delivered through HKPropel that contain 10 interactive lab activities with video to enhance student learning and simulate the experience of performing the labs in the real world A completely new laboratory chapter on high intensity fitness training that includes several popular intermittent fitness tests that students can learn to perform and interpret An appendix that helps estimate the oxygen cost of walking running and cycling New research and information pertaining to each laboratory topic A lab activity finder that makes it easy to locate specific tests In addition to the interactive lab activities which are assignable and trackable by instructors HKPropel also offers students electronic versions of individual and group data sheets of standards and norms question sets to help students better understand laboratory concepts and case studies with answers to further facilitate real world application Chapter quizzes assessments that are automatically graded may also be assigned by instructors to test comprehension of critical concepts Organized in a logical progression the text builds upon the knowledge students acquire as they advance Furthermore the text provides multiple lab activities and includes an equipment list at the beginning of each activity allowing instructors flexibility in choosing the lab activities that will best work in their facility Laboratory Manual for Exercise Physiology Second Edition With HKPropel Access exposes students to a broad

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Laboratory Manual in General Microbiology Michigan Agricultural College. Department of Bacteriology and Hygiene, Michigan State University. Dept. of Bacteriology and Public Health, 1921 The full text of the first edition 1916 is available at <http://www.biodiversitylibrary.org/item/62094>

Physical Laboratory Manual for Use in Schools and Colleges Horatio Nelson Chute, 1895

Resources in Education, 1992-04

Chemical Technology Walter J. Brooking, United States. Division of Vocational and Technical Education, 1964

Collection Management in the Electronic Age Association of College and Research Libraries. Community and Junior College Libraries Section, 1994 Discusses policies for collection management in Learning Resource Centers in the electronic age

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