



Exploring Equilibrium

It Works Both Ways

Introduction

The sound exact librium has two roots segret, meaning recall, and Afric, meaning weight or bulance. Our physical sense of equilibrium—in the notion of a secure or the sering of a perchalance—suggests an equal bulance of appointing forces. How does this physical sense of equilibrium translate to chemical equilibrium? Explore the nature and consequences of equilibrium to the other or a perchalance.

Concepts

Reversible machines.

Chemical equilibrium

· Commercion reaction

Activitude incitorer

Background

Physical changes, such as nielting for or classifying suger, are often increcated by noting that these processes can be easily reversed. Some connum examples of chemical change, such as burning wood or spoiling, fixed, generally cannot be reversed. A closer look at chemical change, however, reveals that, many chemical reactions are also reversible.

Consider the following example of a reversible chemical reaction (Ecuation 1). At high pressures and in the presence of a catalyst, nitrogen and hydrogen react to form ammonia. If the temperature is high enough, however, ammonia decomposes to reform its constituent elements. The traction can go both way? This reversible reaction is represented symbolically using double amount.

Physical Science 2:

What happens when nitrogen and hydrogen are allowed to react? In a closed system, the concentrations of nitrogen and hydrogen will decrease, and the concentrations of anaroma will steadily increase as the maction process in the forward direction. Soon, however, the concentration of anaromic will be large enough that the reverse roution will begin to take place at a significant rate as well. Esembally, as the reaction occurs in both the forward and the research direction, the number of anaromic molecules being formed will become estad to the number of anaromic numbersheets being consumed. At this point, no further changes a will be observed in the overall concentrations of nitrogen, hydrogen and a nonconta. This is the point of chemical equilibrium.

Observined equilibrium is defined, as the state in which the rate of the forward reaction equals the rate of the reverse reaction and the concerntrations of reactants and products remain constant, with time. Note that, this definition describes a dynamic picture of equilibrium. The reactions continue, but there is an equal balance of opposing reaction rates.

What happens when the southinions is disperbed? Any factor that changes the rate of the forward or the reverse reaction will change the associate of reactions and products that are present at equilibrium. Reaction conditions that are known to allow the rates of chemical reactions include the concentrations of reactions and the temperature. In this experiment, we will investigate how changes in traction conditions affect the associates of reactions are therefore a conditions of the associates of reactions are products present at equilibrium.

Experiment Overview

The purpose of this experiment is to explore the nature of chemical equilibrium and unitiaritify conditions that affect the position of equilibrium. Two different reversible reactions will be studied.

Reaction of iron(III) nitrate with potassium thiocyanate will be used to study complie interpretarion. Iron(III) sons reset with thiocyanate torse to form PeSCN¹ complex ions (Equation 2). The offects of changing the concentrations of reactions and of changing the resetation tentralizes will be investigated.

Planation 2

The properties of an inclinator will be used to study and Autoropoliteiston. An inclinator is a dynthot one gain or lose hydrogen ions to force substances that have different orders. Equation 3 summarises the reservible reaction of the inclinator branches of green (HIrd). His represents an uncharged inclinator molecule and. In an indicator arises formed after the molecule has lost a hydrogen ion. The color of the indicator in the presence of either reacts H. or CSH, ions (see Equation 4) will show have changing the concentration of a problem, affects the equilibrium shows in Equation 3.

Cartine To.

Physics Phone 3

 $H(ac) + OH(ac) \rightarrow H_1O0$

Elpaintion 4

<u>Lab Report On Exploring Equilibrium It Works Both</u> <u>Ways</u>

RJ Shavelson

Lab Report On Exploring Equilibrium It Works Both Ways:

Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course. The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science Yucca Mountain Site Characterization Project Bibliography, Scientific and Technical Aerospace Reports, 1994 Lists citations with abstracts for aerospace related 1994-1995 ,1996 reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database Research and Development in Progress U.S. Atomic Energy Commission. **Energy Research Abstracts** ,1993 Government Reports Announcements Division of Biology and Medicine, 1968 History of Science in United States Marc Rothenberg, 2012-10-12 This Encyclopedia examines all aspects of the .1974 history of science in the United States with a special emphasis placed on the historiography of science in America It can be used by students general readers scientists or anyone interested in the facts relating to the development of science in the United States Special emphasis is placed in the history of medicine and technology and on the relationship between science and technology and science and medicine Cities and Their Vital Systems Advisory Committee on Technology and Society, 1989 Cities and Their Vital Systems asks basic questions about the longevity utility and nature of urban infrastructures analyzes how they grow interact and change and asks how when and at what cost they should be replaced Among the topics discussed are problems arising from increasing air travel and airport congestion the adequacy of water supplies and waste treatment the impact of new technologies on construction urban real estate values and the field of telematics the combination of computers and telecommunications that makes money machines and national newspapers Job Creation and Local Economic Development OECD, 2014-11-19 This publication highlights new evidence on possible policies to support job creation bringing together the latest research on labour market entrepreneurship and local economic development policy to help governments support job creation in the recovery Routledge Handbook of Energy Economics Uğur Soytaş, Ramazan Sarı, 2019-09-23 Energy consumption and production have major influences on the economy environment and society but in return they are also influenced by how the economy is structured how the social institutions work and how the society deals with environmental degradation The need for integrated assessment of the relationship between energy economy environment and society is clear and this handbook offers an in depth review of all four pillars of the energy economy environment society nexus Bringing together contributions from all over the world this handbook

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United States. Energy Research and Development Administration, 1977-03

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