Chem lab report

Design

Aim: To determine the moles of water of crystallization in oxalic acid.

Hypothesis: If NaOH is added to Hydrated Oxalic acid, then the hydrogen ions from the acid will react with hydroxide ions from NaOH to produce water. The rest forms a colorless salt.

In this experiment, the moles of water of crystallization is calculated using titration. An indicator is used to show when the titration has ended. Since two moles of NaOH react with one mole of oxalic acid, we can find out the moles of oxalic acid used, then using the mass of the crystals taken, find the mass of water present in the crystals which will give the molecules of water of crystallization in oxalic acid.

Variables:

- Independent :
 - Volume of NaOH used- since 25cm³ of NaOH is taken for all trials
- · Dependent:
 - Volume of oxalic acid solution used-the volume of oxalic acid solution used would

Lab 57 Titration Oxalic Acid

Jessica J Manson

Lab 57 Titration Oxalic Acid:

Unveiling the Magic of Words: A Overview of "Lab 57 Titration Oxalic Acid"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "Lab 57 Titration Oxalic Acid," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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