

The Chemistry of Life ANSWER KEY

Section 2-1 The Nature of Matter (pages 35-39)

This section identifies the three particles that make up atoms. It also explains how atoms of the same element can have a different number of neutrons and describes the two main types of chemical bonds.

Atoms (page 35)

1. The basic unit of matter is called a(an) atom.
It is the center of an atom, made up of protons and neutrons.
2. Describe the nucleus of an atom.

3. Complete the table about subatomic particles.

SUBATOMIC PARTICLES

Particle Charge Location in Atom

Proton - Positive - Nucleus

Neutron - Neutral - Nucleus

Electron - Negative- Surrounding nucleus

4. Why are atoms neutral despite having charged particles?

Atoms have equal numbers of electrons and protons, and these subatomic particles have equal, but opposite, charges.

Elements and Isotopes (page 36)

5. What is a chemical element?

A chemical element is a pure substance that consists entirely of one type of atom.

6. What does an element's atomic number represent?

It represents the number of protons in an atom of the element.

7. Atoms of the same element that differ in the number of neutrons they contain are known as isotopes.

8. How are isotopes identified? Isotopes are identified by their mass number.

9. Why do all isotopes of an element have the same chemical properties? They have the same chemical properties because they have the same number of electrons.

Chemical Compounds (page 37)

10. What is a chemical compound?

A chemical compound is a substance formed by the chemical combination of two or more elements in definite proportions.

11. What does the formula for table salt indicate about that compound? The formula for table salt, NaCl, indicates that the elements from which table salt forms—sodium and chlorine—combine in a 1:1 ratio.

Chemical Bonds (pages 38-39)

Lab 5 Chemistry Of Life

Julia Schneider



Lab 5 Chemistry Of Life:

TEACHING OF BIOLOGICAL SCIENCES (Intended for Teaching of Life Sciences, Physics, Chemistry and General Science) AHMAD, JASIM, 2011-11-30 Students of today especially at the school level perceive science as a collection of facts to be memorized whereas in reality it is constantly changing as new information accumulates and new techniques develop every day The objective of teaching is not restricted to imparting scientific information to students but also to help them apply these principles in their daily lives This comprehensive book written in an easy to understand language covers the entire syllabus of teaching of Biological Sciences in particular and Science Teaching in general In so doing it takes into account the needs of teacher trainees and in service teachers Organized into 20 chapters the book discusses in detail the many facets and aspects of Biology Science Teaching The text introduces modern approaches to teaching with the aim of improving student learning throughout their course It emphasizes the need for pedagogical analysis vis vis subject teaching constructive approach laboratory work Continuous and Comprehensive Evaluation CCE In addition the text highlights the difference between microteaching and simulated teaching It also shows how e learning and co curricular activities can be successfully integrated in biological sciences teaching NEW TO THIS EDITION Inclusion of one chapter on Concept Mapping in Biology Teaching This chapter advocates the popularized constructivist approach of teaching learning process Besides some figures tables and flow charts are also added to make the book more useful to the readers KEY FEATURES Analyses Constructivism versus Behaviourism Includes self explanatory model lesson plan Discusses Information and Communication Technology ICT in the context of Biology Science teaching learning Suggests how apparatus and devices can be secured and cultured and used in classroom demonstrations and student projects Primarily intended as a text for students of B Ed pursuing course on Teaching of Biological Sciences Life Sciences the book should prove equally useful for B Ed students following courses on Teaching of Physical Sciences In addition diploma students of Elementary Teacher Education ETE having a paper on Teaching of EVS General Science and M Ed and M A Education students with an optional elective paper on Science Education would find the book extremely useful

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University in Wales After a postdoctoral stay with Kaoru Fuji at Kyoto University as a JSPS fellow he started his independent research in the group of Bernd Giese in Basel Switzerland He was invited as a visiting professor to various places University of Toronto Canada 1999 Chuo University in Tokyo Japan 2000 Osaka University Japan 2004 He was awarded the Werner Prize from the New Swiss Chemical Society in 2000 He is the author of about 80 publications and has written or edited 4 books *the hawaiian forester an dagriculturist* leopold g. blackman,1908 *The Hawaiian Forester and Agriculturist* ,1908 *Applications of Mass Spectrometry in Life Safety* Crisan Popescu,Alina D. Zamfir,Nicolae Dinca,2008-09-02 Mass spectrometry MS along with its hyphenated techniques is capable of high throughput sensitivity accuracy and selectivity for the analysis of structure and composition of almost any product Like in electrophoresis MS separates molecules based on the mass to charge ratio In case of gel electrophoresis SDS PAGE a well known and efficient bioanalytical technique proteins bear negative charges but have the same charge density so proteins are separated according to their size Similarly in case of MS analysis proteins carry the same charge and are separated by their molecular weight Unlike SDS PAGE however modern ultra high resolution MS discerns very small mass differences and can resolve and completely identify in a single experiment species of the same nominal mass in complex biological mixtures Consequently MS can be used for the structural characterization identification and sensitive detection of mixtures of biomolecules or for assessing the quality of isolated proteins purity integrity or post translational modifications for example carbohydrates nucleic acids drugs metabolites pollutants etc In the post genome era MS is continuously developing as one of the most reliable analytical method for elucidating the structure of molecules originating from various biological matrices The potential of MS for high sensitive structural analyses became unsurpassable after the introduction of electrospray ESI and matrix assisted laser desorption ionization MALDI methods on one hand and the possibility to deduce in detail unknown biopolymer structures by highly accurate molecular mass measurement followed by sequencing using dissociation techniques based on multiple stage MS on the other

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