



Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

Assessment Statements

- | | |
|--|---|
| <ul style="list-style-type: none"> <input type="checkbox"/> Outline the Scientific Method <input type="checkbox"/> Identify independent, dependent & controlled variables. <input type="checkbox"/> Demonstrate correct and safe use of laboratory equipment. <input type="checkbox"/> Distinguish between quantitative and qualitative data <ul style="list-style-type: none"> ➢ <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i> ➢ <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> ◊ <i>Raw data (as recorded)</i> ◊ <i>Processed (after calculations)</i> <input type="checkbox"/> Distinguish between accuracy and precision <ul style="list-style-type: none"> ➢ <i>Accuracy describes the correctness of a set of measurements</i> ➢ <i>Precision describes the repeatability of a set of measurements</i> | <ul style="list-style-type: none"> <input type="checkbox"/> State SI units of measurement for <ul style="list-style-type: none"> ➢ <i>Mass: grams (g)</i> ➢ <i>Length: Metres (m), millimeters (mm)</i> ➢ <i>Concentration: molarity (M)</i> ➢ <i>Temperature: Degrees Celsius (°C)</i> ➢ <i>Volume: Litres (L) or millilitres (mL)</i> ➢ <i>Density: grams per cubic centimeter (g/cm³)</i> ➢ <i>Conductivity: Siemens (S)</i> ➢ <i>Acidity or alkalinity (pH)</i> <input type="checkbox"/> Energy: Joules (J or kJ)Determine the uncertainty of digital and analog measuring tools <ul style="list-style-type: none"> ➢ <i>Usually ± 1 of the smallest division</i> <input type="checkbox"/> Calculate the % error of a given or recorded set of values <ul style="list-style-type: none"> ➢ <i>$(\text{Measured} - \text{True}) / \text{True} \times 100$</i> <input type="checkbox"/> State values to appropriate numbers of significant figures <input type="checkbox"/> Calculate values to appropriate numbers of significant figures |
|--|---|

Related Concepts:

- ☐ 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm#>)
- ☐ 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm#>)
- ☐ 01.3 SigDigs and Notation (Quia quiz: <http://www.quia.com/quiz/3504378.htm#>)

Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

Apply	Use knowledge & understanding in response to a real situation or given circumstances.
Calculate	Obtain a numerical answer, showing relevant stages in the working.
Define	Give the precise meaning of a word, phrase, concept or physical quantity.
Describe	Give a detailed account or picture of an event, pattern, process or pattern.
Design	Produce a plan, simulation or model.
Determine	Obtain the only possible answer.
Discuss	Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence.
Distinguish	Make clear the differences between two or more concepts or items.
Evaluate	Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.
Explain	Give a detailed account including reasons or causes.
Identify	Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.
Label	Add title, labels or brief explanation(s) to a diagram or graph.
List	Give a sequence of answers with no explanation.
Outline	Give a brief account.
Recall	Remember or recognize from prior learning experiences.
State	Give a specific name, value or other brief answer without explanation or calculation.
Suggest/Propose	Propose a solution, hypothesis or other possible answer.

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order **Basic Concepts of Chemistry** Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance **Ency. Of Teaching Of Science 2Vol.** Poly Dat, 2004 **The Annual American Catalog** ,1904 *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office, 1968 **The Public Library Quarterly** ,1902 **Teaching and Learning in the School Chemistry Laboratory** Avi Hofstein, Muhamad Hugerat, 2021-11-26 Authored by renowned experts in the field of chemistry education this book provides a holistic approach to cover all issues related to learning and teaching in the chemistry laboratory English Communication for International Teaching Assistants Greta Gorsuch, Colleen M. Meyers, Lucy Pickering, Dale T. Griffiee, 2016-08-26 In order to become successful communicators in undergraduate teaching environments in the United States nonnative English speaking graduate students require hands on and relevant practice Now in its second edition English Communication for International Teaching Assistants continues to provide constructive insights and versatile tools to facilitate such learning The authors convey the principals of cross cultural awareness within the classroom and emphasize the importance of discourse intonation in building rapport conveying meaning and structuring

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