

Lab 008 – pH and Buffers

A. Objectives:

1. Constructive proper titration and titration.
2. Construct and explain how the pH scale works.
3. Construct and explain the concept of acids, bases, and buffers.
4. Construct acids and bases in the laboratory.
5. Calculate the pH of various acids/bases.
6. Calculate equilibrium pH, buffer, and buffer capacity of acids and bases.

B. Scientific Method: Data Tables and Graphs

1. Data Tables

Construct a table in your laboratory notebook that will allow you to record the following data: pH, buffer capacity, buffer, and buffer capacity. (Construct a table below.)

2. Construct and explain the pH scale. (Construct a table in your laboratory notebook that will allow you to record the following data: pH, buffer capacity, buffer, and buffer capacity. (Construct a table below.)

3. Construct the equilibrium constant expression for the acid dissociation reaction. (Construct a table in your laboratory notebook that will allow you to record the following data: pH, buffer capacity, buffer, and buffer capacity. (Construct a table below.)

4. Construct the equilibrium constant expression for the acid dissociation reaction. (Construct a table in your laboratory notebook that will allow you to record the following data: pH, buffer capacity, buffer, and buffer capacity. (Construct a table below.)

Equilibrium Constant	Equilibrium	Equilibrium	Equilibrium	Equilibrium	Equilibrium
Equilibrium Constant	Equilibrium	Equilibrium	Equilibrium	Equilibrium	Equilibrium
Equilibrium Constant	Equilibrium	Equilibrium	Equilibrium	Equilibrium	Equilibrium
Equilibrium Constant	Equilibrium	Equilibrium	Equilibrium	Equilibrium	Equilibrium

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