

KENMORE INDUCTION COOKTOP INSTALLATION INSTRUCTIONS

INSTALLATION AND SERVICE MUST BE PERFORMED BY A QUALIFIED INSTALLER.

IMPORTANT: SAVE FOR LOCAL ELECTRICAL INSPECTOR'S USE.

READ AND SAVE THESE INSTRUCTIONS FOR FUTURE REFERENCE.

⚠ WARNING FOR YOUR SAFETY: Do not store or use gasoline or other flammable vapors and liquids in the vicinity of this or any other appliance.

IMPORTANT INSTALLATION INFORMATION

- All electric cooktops run off a single phase, three-wire or four-wire cable, 240/208 volt, 60 hertz, AC only electrical supply with ground.
- Minimum distance between cooktop and overhead cabinetry is 30" (76.2 cm).

* 30" (76.2 cm) min. for unprotected cabinet and 24" (61 cm) min. for cabinet with protected bottom surface.

** DO NOT obstruct these areas.

DO NOT apply any sealant along the perimeter. The cooktop must breathe freely.



Cooktop Dimensions

Cooktop Cutout Dimensions

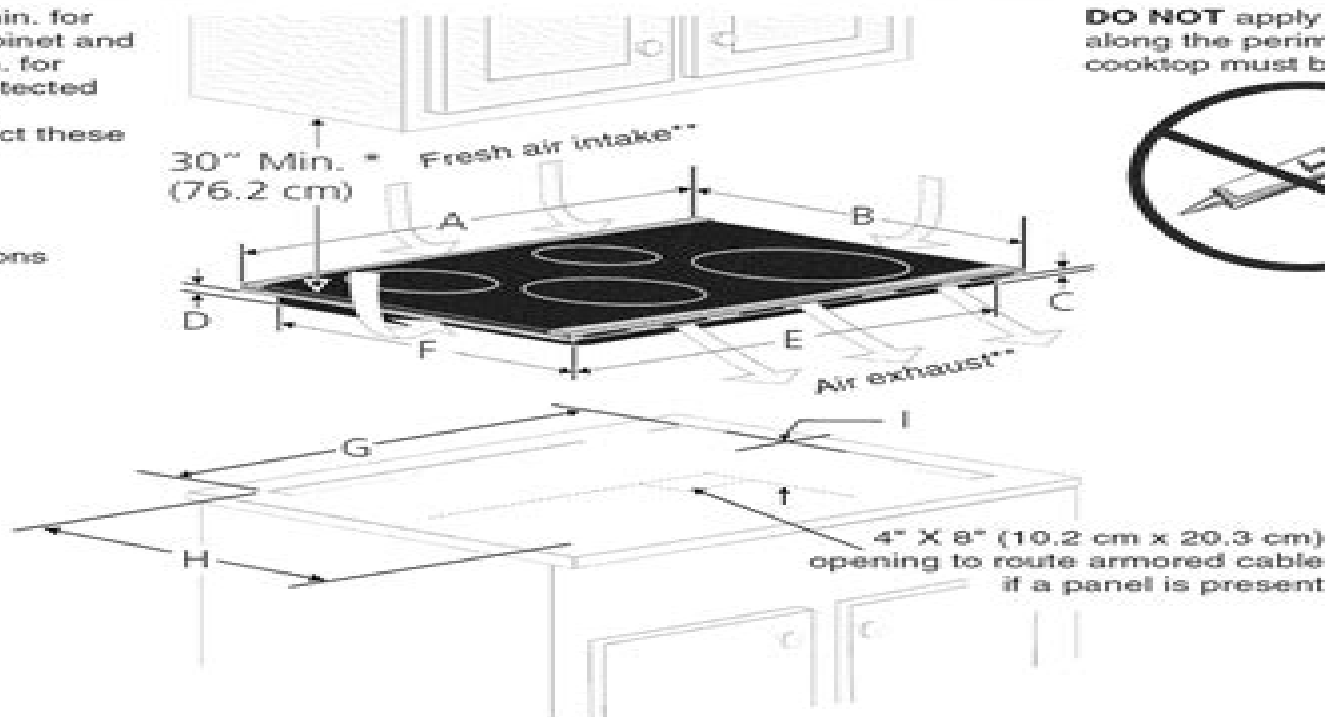


Figure 1

		COOKTOP DIMENSIONS				CUTOUT DIMENSIONS					
A. WIDTH	B. DEPTH	HEIGHT		E. WIDTH	F. DEPTH	G. WIDTH		H. DEPTH		I. HEIGHT	
		C	D			Min.	Max.	Min.	Max.		
30 3/4 (78.1)	21 1/2 (54.6)	3 (7.6)	4 3/4 (11.1)	28 1/4 (73.3)	20 (50.8)	29 3/4 (75.2)	29 3/4 (75.6)	20 3/4 (51.8)	20 3/4 (52.1)	4 1/2 (11.4)	
36 1/4 (93.3)	21 1/2 (54.6)	3 (7.6)	4 3/4 (11.1)	35 1/4 (89.5)	20 (50.8)	35 3/4 (91.1)	36 (91.4)	20 3/4 (51.8)	20 3/4 (52.1)	4 1/2 (11.4)	

All dimensions are stated in inches and (cm).

Allow 2" (5 cm) space below the armored cable opening to clear the electric cable and allow space for installation of the junction box on the wall at the back of the cooktop.

P/N 318205407 (0807) Rev. A

English – pages 1-6

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Consumer Reports



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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web lab 4 stoichiometry the reaction of iron with copper ii sulfate revised 8 19 2009 1 introduction in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of

copper ii sulfate this reaction produces

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated CuSO_4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry the reaction of iron with copper ii sulfate purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g