

# IGCSE Physics

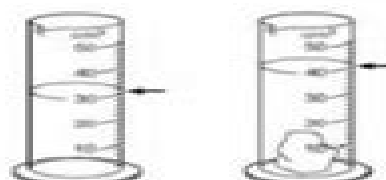
## Volume, Density and Mass

**Volume ( $\text{cm}^3$ )** – the amount of space an object occupies

How to measure volume:

- **For Liquids** – use measuring cylinder and read the volume directly
- **For Regular Solids** – calculate using the objects measurements:
  - Cuboid: height  $\times$  width  $\times$  length
  - Sphere:  $\frac{4}{3} \times \pi \times \text{radius}^3$ .
  - Cylinder:  $\pi \times \text{radius}^2 \times \text{height}$
- **For Irregular Solids** – use the displacement method

**Displacement method** – place a selection of water in a measuring cylinder and measure the volume. Then add the object into the water and measure it again. The volume of the object is the difference in volumes. *E.g.*



Initial Volume =  $30\text{cm}^3$   
 Final Volume =  $40\text{cm}^3$   
 $40\text{cm}^3 - 30\text{cm}^3 = 10\text{cm}^3$   
 Object Volume =  $10\text{cm}^3$

**Density ( $\text{g/cm}^3$ )** – mass per unit of volume

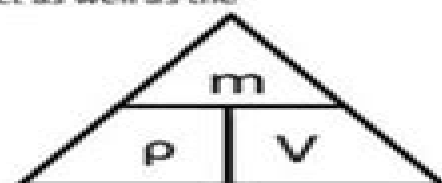
**Relative Density** – the ratio between the density of that substance and the density of water. It has no limit.

**Mass (g)** – a measure of the amount of matter present in an object as well as the characteristic which resists change in motion

The calculation of volume is  $\text{volume} = \frac{\text{mass}}{\text{density}}$

The calculation of density is  $\text{density} = \frac{\text{mass}}{\text{volume}}$

The calculation of mass is  $\text{mass} = \text{density} \times \text{volume}$



**Density of air** –  $1.225 \text{ kg/m}^3$

**Density of water** –  $1000 \text{ kg/m}^3$

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**Graham Booth**



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